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APPENDIX C
Atlas of
Innershell Ionization lines

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APPENDIX C

INNERSHELL IONIZATION LINES

1 Introduction

In this Appendix we present an extended line table (Table 1) which lists more than 400 $K\alpha$ and $K\beta$ fluorescence lines of elements with $Z = 6-28$ compiled from a recent publication (KM93). In the following chapters we give first the formula as derived by MG81 for the innershell ionization contribution to a line in an optically thin plasma (§2), then consider the various literature sources and the way we derive wavelengths, and finally give the line table.

2 Formula for innershell ionization

Mewe and Gronenschild (MG81) and Mewe *et al.* (MSS80) have considered electron impact ionization from the inner $1s^2$ shell of ion $Z^{(z-1)}$ which contributes to the production in an optically thin plasma to the formation of a certain line in the next higher ion Z^{+z} (e.g. the satellites to He-like resonance lines from lower Li-, Be-, etc. like ions). On the basis of the formula of Lotz (L68) with $\zeta = 2$ (total number of electrons in the $1s^2$ shell) and assuming the upper line level to be populated in proportion to its statistical weight g_j MG81 obtain for the rate coefficient:

$$S_{II} = 6.49 \times 10^{-4} C_{II} T^{-1/2} \chi^{-1} E_1(\chi/kT) \text{ (cm}^3\text{s}^{-1}\text{)}, \quad (1)$$

where the fluorescence yield is given by

$$C_{II} = \frac{g_j}{\sum g_j} \frac{A_r}{A_a + \sum A_r}. \quad (1a)$$

Here $\sum g_j$ is the total statistical weight of the configuration to which level j belongs and A_a, A_r are transition probabilities for decay by auto-ionization and radiation, where the summation is over all possible spontaneous radiative transitions from level j . The electron temperature T is here expressed in K and the ionization energy χ of the $1s^2$ shell in ion $Z^{+(z-1)}$ is expressed in eV (cf. Eq. (21a) of chapter 'LINEM').

In recent work Kaastra and Mewe (KM93) have calculated the yield C_{II} for the $K\alpha$ and $K\beta$ lines from many ionization stages of the cosmically abundant elements. These are given in Table 1 and will replace the previously calculated yields given by MSS80 and MG81.

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3 Literature sources and wavelengths

3.1 $K\alpha$ transitions

For the $K\alpha$ transitions the energy and wavelength in the second and third column of Table 1 are derived by KM93 based on data of L68, whereas the (slightly more accurate) energy and wavelength in the last two columns are the data from H69 corrected by a comparison with the data from BB67 for $K\alpha$ transitions in singly ionized atoms. In the latter case, we apply, for example for iron and calcium for all ionization stages a shift of -0.017 \AA , i.e. we use the 'adjusted' values given by H69.

For $Z > 13$ we take for the most abundant elements (Si, Ca, Fe) a doublet splitting equal to that derived from the BB67 data for singly ionized atoms (e.g., 0.0024, 0033, and 0.0039 \AA for SiII, CaII, and FeII, respectively). We apply this splitting only up to the stage where the $2p$ electron jumps from a full $2p$ shell (e.g. up to Fe XVIII where L68 still gives different energies for the $K\alpha_1$ (Kal1) and $K\alpha_2$ (Kal2) components), thereafter we give the average-of-configuration data from H69. For $Z \leq 13$ we give the averaged data for all more than singly ionized atoms. We note that the H69 data are in good agreement with those from MW86 calculated for $Z = 9$ to 18.

3.2 $K\beta$ transitions

For the $K\beta$ lines we take the data from BB67 for singly ionized atoms and in the case of titanium from M83 for the higher ions. Corresponding values for other ions we approximately derived by scaling λ_H (lam_H) for $K\alpha_2$ (Kal2) with the ratio of the wavelengths of $K\beta_3$ (Kbe3) and $K\alpha_2$ (Kal2) from KM93 which are based on the data from L68. For the two components $K\beta_3$ (Kbe3) and $K\beta_1$ (Kbe1) we take the same wavelength except for iron where we apply a splitting Kbe3-Kbe1 of 0.0004 \AA based on the L68 data up to the stage Fe X where the $3p$ electron still jumps from a full $3p$ shell and L68 gives different energy values for the two components.

4 Remarks and explanation of Table 1

To convert energies into wavelengths and vice versa we use a conversion factor of $12.39842 \text{ \AA} / \text{keV}$.

The energy and wavelength should be taken from the last two columns of Table 1 (i.e. the values for E_H and lam_H based on BB67 and H69), and not those from the second and third column (values for E and λ from KM93 based on L68).

Explanation of Table 1:

Z = atomic number, s = ionization stage, l = line transition (notation from KM93), E = energy, λ (lambda) = wavelength (from KM93), yield (= C_{II}) = fluorescence yield (photons per ionization) (from KM93), transition = ion and line transition, E_H (E_H) = energy, λ_H (lam_H) = wavelength (from H69 and BB67) .

5 References

- Bearden, J.A., Burr, A.F.: 1967, *Rev. Mod. Phys.* **39**, 125 (**BB67**)
House, L.L.: 1969, *Astrophys. J. Suppl.* **18**, 21 (**H69**)
Kaastra, J.S., Mewe, R.: 1992, *Astron. Astrophys. Suppl.* , **443** (**KM93**)
Lotz, W.: 1968, *Journ. Opt. Soc. Am.* **58**, 915 (**L68**)
Maurer, R.J., Watson, R.L.: 1986, *At. Data & Nucl. Data Tables* **34**, 186 (**MW86**)
Mewe, R., Gronenschild, E.H.B.M.: 1981, *Astron. Astrophys. Suppl. Ser.* **45**, 11 (**MG81**)
Mewe, R., Schrijver, J., Sylwester, J.: 1980, *Astron. Astrophys. Suppl. Ser.* **40**, 323 (**MSS80**)
Morita, S. *et al.*: 1983, *Phys. Letters* **94A**, 147 (**M83**)

Table 1. List of $K\alpha$ and $K\beta$ innershell ionization lines (continued)

29 1 1	8.0279	1.5444	0.1311	CuII	Kal2	8.0278	30 2 3	9.5896	1.2929	0.0168
29 1 2	8.0482	1.5405	0.2621	CuII	Kal1	8.0478	30 2 4	9.5940	1.2923	0.0336
29 1 3	8.9023	1.3927	0.0166	CuII	Kbe3	8.9029	30 3 1	8.6397	1.4350	0.1318
29 1 4	8.9064	1.3921	0.0332	CuII	Kbel	8.9053	30 3 2	8.6636	1.4311	0.2635
29 2 1	8.0408	1.5419	0.1312				30 3 3	9.6139	1.2896	0.0168
29 2 2	8.0593	1.5384	0.2623				30 3 4	9.6183	1.2890	0.0336
29 2 3	8.9248	1.3892	0.0166				30 4 1	8.6537	1.4327	0.1319
29 2 4	8.9289	1.3886	0.0333				30 4 2	8.6776	1.4288	0.2637
29 3 1	8.0538	1.5395	0.1313				30 4 3	9.6383	1.2864	0.0168
29 3 2	8.0742	1.5356	0.2625				30 4 4	9.6427	1.2858	0.0336
29 3 3	8.9516	1.3851	0.0166				30 5 1	8.6576	1.4304	0.1320
29 3 4	8.9536	1.3847	0.0333				30 5 2	8.6916	1.4265	0.2639
29 4 1	8.0686	1.5366	0.1314				30 5 3	9.6650	1.2828	0.0168
29 4 2	8.0872	1.5331	0.2627				30 5 4	9.6694	1.2822	0.0337
29 4 3	8.9764	1.3812	0.0166				30 6 1	8.6816	1.4281	0.1321
29 4 4	8.9784	1.3809	0.0333				30 6 2	8.7056	1.4242	0.2641
29 5 1	8.0816	1.5341	0.1315				30 6 3	9.6895	1.2796	0.0168
29 5 2	8.1021	1.5303	0.2629				30 6 4	9.6939	1.2790	0.0337
29 5 3	9.0012	1.3774	0.0167				30 7 1	8.6976	1.4255	0.1322
29 5 4	9.0033	1.3771	0.0333				30 7 2	8.7197	1.4219	0.2643
29 6 1	8.0965	1.5313	0.1315				30 7 3	9.7163	1.2760	0.0169
29 6 2	8.1171	1.5274	0.2631				30 7 4	9.7208	1.2755	0.0337
29 6 3	9.0261	1.3736	0.0167				30 8 1	8.7116	1.4232	0.1323
29 6 4	9.0303	1.3730	0.0333				30 8 2	8.7357	1.4193	0.2645
29 7 1	8.1133	1.5282	0.1316				30 8 3	9.7432	1.2725	0.0169
29 7 2	8.1339	1.5243	0.2633				30 8 4	9.7477	1.2719	0.0337
29 7 3	9.0552	1.3692	0.0167				30 9 1	8.7257	1.4209	0.1324
29 7 4	9.0573	1.3689	0.0334				30 9 2	8.7498	1.4170	0.2647
29 8 1	8.1246	1.5260	0.1317				30 9 3	9.7701	1.2690	0.0169
29 8 2	8.1452	1.5222	0.2635				30 9 4	9.7746	1.2684	0.0338
29 8 3	9.0782	1.3657	0.0167				3010 1	8.7398	1.4186	0.1325
29 8 4	9.0803	1.3654	0.0334				3010 2	8.7640	1.4147	0.2649
29 9 1	8.1395	1.5232	0.1318				3010 3	9.7949	1.2658	0.0169
29 9 2	8.1583	1.5197	0.2637				3010 4	9.7994	1.2652	0.0338
29 9 3	9.1033	1.3620	0.0167				3011 1	8.7539	1.4163	0.1326
29 9 4	9.1075	1.3613	0.0334				3011 2	8.7781	1.4124	0.2651
2910 1	8.1545	1.5204	0.1319				3011 3	9.8220	1.2623	0.0169
2910 2	8.1733	1.5169	0.2639				3011 4	9.8265	1.2617	0.0338
2910 3	9.1285	1.3582	0.0167				3012 1	8.7700	1.4137	0.1327
2910 4	9.1327	1.3576	0.0334				3012 2	8.7922	1.4102	0.2653
2911 1	8.1696	1.5176	0.1320				3012 3	9.8492	1.2588	0.0169
2911 2	8.1884	1.5141	0.2641				3012 4	9.8537	1.2582	0.0338
2911 3	9.1538	1.3545	0.0167				3013 1	8.7842	1.4115	0.1328
2911 4	9.1580	1.3538	0.0335				3013 2	8.8064	1.4079	0.2655
2912 1	8.1828	1.5152	0.1321				3013 3	9.8742	1.2556	0.0169
2912 2	8.2035	1.5114	0.2643				3013 4	9.8787	1.2551	0.0339
2912 3	9.1791	1.3507	0.0167				3014 1	8.8004	1.4089	0.1361
2912 4	9.1833	1.3501	0.0335				3014 2	8.8227	1.4053	0.2722
2913 1	8.1979	1.5124	0.1354				3014 3	9.9038	1.2519	0.0174
2913 2	8.2186	1.5086	0.2708				3014 4	9.9038	1.2519	0.0260
2913 3	9.2087	1.3464	0.0172				3015 1	8.8125	1.4069	0.1395
2913 4	9.2087	1.3464	0.0257				3015 2	8.8369	1.4030	0.2791
2914 1	8.2130	1.5096	0.1388				3015 3	9.9243	1.2493	0.0178
2914 2	8.2338	1.5058	0.2777				3015 4	9.9243	1.2493	0.0178
2914 3	9.2300	1.3433	0.0176				3016 1	8.8267	1.4046	0.1431
2914 4	9.2300	1.3433	0.0176				3016 2	8.8512	1.4008	0.2863
2915 1	8.2262	1.5072	0.1424				3016 3	9.9312	1.2484	0.0183
2915 2	8.2471	1.5034	0.2848				3016 4	9.9312	1.2484	0.0091
2915 3	9.2385	1.3420	0.0181				3017 1	8.8410	1.4024	0.1469
2915 4	9.2385	1.3420	0.0090				3017 2	8.8654	1.3985	0.2938
2916 1	8.2414	1.5044	0.1461				3017 3	9.9541	1.2456	0.0187
2916 2	8.2604	1.5010	0.2923				3018 1	8.8552	1.4001	0.1507
2916 3	9.2598	1.3390	0.0185				3018 2	8.8797	1.3963	0.3015
2917 1	8.2547	1.5020	0.1500				3018 3	9.9724	1.2433	0.0096
2917 2	8.2756	1.4982	0.2999				3019 1	8.8675	1.3982	0.1548
2917 3	9.2790	1.3362	0.0095				3019 2	8.8920	1.3943	0.3096
2918 1	8.2680	1.4996	0.1540				3020 1	8.8797	1.3963	0.1567
2918 2	8.2871	1.4961	0.3079				3020 2	8.9043	1.3924	0.3134
2919 1	8.2794	1.4975	0.1559				3021 1	8.8941	1.3940	0.1586
2919 2	8.3004	1.4937	0.3118				3021 2	8.9187	1.3902	0.3172
2920 1	8.2909	1.4954	0.1578				3022 1	8.9207	1.3898	0.2045
2920 2	8.3119	1.4916	0.3156				3022 2	8.9207	1.3898	0.3067
2921 1	8.3157	1.4910	0.2034				3023 1	8.9166	1.3905	0.2766
2921 2	8.3157	1.4910	0.3051				3023 2	8.9166	1.3905	0.2766
2922 1	8.3138	1.4913	0.2750				3024 1	8.8818	1.3959	0.4033
2922 2	8.3138	1.4913	0.2750				3024 2	8.8818	1.3959	0.2016
2923 1	8.2813	1.4972	0.4005				3025 1	8.8695	1.3979	0.6742
2923 2	8.2813	1.4972	0.2002				3026 1	8.8512	1.4008	0.6322
2924 1	8.2699	1.4992	0.6678							
2925 1	8.2547	1.5020	0.6215							
30 1 1	8.6159	1.4390	0.1316	ZnII	Kal2	8.6158				
30 1 2	8.6397	1.4350	0.2631	ZnII	Kal1	8.6389				
30 1 3	9.5675	1.2959	0.0168	ZnII	Kbe3	9.5720				
30 1 4	9.5719	1.2953	0.0335	ZnII	Kbel	9.5720				
30 2 1	8.6258	1.4374	0.1317							
30 2 2	8.6497	1.4334	0.2633							

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 N.B.
 Explanation of the table:
 Z = atomic number, s = ionization stage, l = line transition
 (notation from KM92), E = energy, lambda = wavelength (from KM92)
 yield = fluorescence yield (photons per ionization) (from KM92),
 transition = ion and line transition, E_H = energy, lam_H =
 wavelength (from H69 and BB67).